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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.037$
$w R$ factor $=0.104$
Data-to-parameter ratio $=13.3$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 2-(3-Benzoylthioureido) propionic acid

The molecular structure of the title compound, $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$, adopts a cis-trans configuration with respect to the positions of the propionic acid and benzoyl groups relative to the S atom across the thiourea $\mathrm{C}-\mathrm{N}$ bonds. In the crystal structure, molecules are linked by $\mathrm{O}-\mathrm{H} \cdots \mathrm{S}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions, forming a one-dimensional chain parallel to the $b$ axis.

## Comment

The title compound, (I), is isomorphous with 3-(3-benzoylthioureido)propionic acid (Yusof \& Yamin, 2003); they differ in that the 3-benzoylthioureido group is attached to the $\beta$ alanine group in the latter compound. The molecular structure adopts a cis-trans configuration with respect to the positions of the propionic acid and benzoyl groups relative to the S atom across the $\mathrm{C} 8-\mathrm{N} 2$ and $\mathrm{C} 8-\mathrm{N} 1$ bonds, respectively.

(I)

The bond lengths and angles (Table 1) are in normal ranges (Allen et al.,1987) and comparable with those in the $\beta$-alanine derivative. The central carbonylthiourea group (S1/C8/N1/N2/ C 7 ), and the phenyl ( $\mathrm{C} 1-\mathrm{C} 6$ ) and ethanoic acid fragments are planar, with a maximum deviation of 0.037 (2) A for atom C7. The central carbonylthiourea group makes dihedral angles


Figure 1
The molecular structure of the title compound, (I), with displacement ellipsoids drawn at the $50 \%$ probability level. Dashed lines indicate hydrogen bonds.

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Figure 2
Packing diagram of compound (I), viewed down the $c$ axis. The dashed lines denote $\mathrm{O}-\mathrm{H} \cdots \mathrm{S}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds.
with the phenyl and ethanoic acid fragments of 27.52 (9) and $24.20(12)^{\circ}$, respectively. The phenyl ring is inclined to the ethanoic acid fragment by 4.19 (4) ${ }^{\circ}$.

There are three intramolecular hydrogen bonds C9$\mathrm{H} 9 \cdots \mathrm{~S} 1, \mathrm{~N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 1$ and $\mathrm{N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 3$ (Table 2) and, as a result, two pseudo-five-membered rings ( $\mathrm{C} 9-\mathrm{H} 9-\mathrm{S} 1-$ $\mathrm{C} 8-\mathrm{N} 2)$, ( $\mathrm{N} 2-\mathrm{H} 2 A-\mathrm{O} 3-\mathrm{C} 11-\mathrm{C} 9$ ) and a pseudo-sixmembered ring ( $\mathrm{N} 2-\mathrm{H} 2 A-\mathrm{O} 1-\mathrm{C} 7-\mathrm{N} 1-\mathrm{C} 8$ ) are formed (Fig. 1). In the crystal structure, the molecules are linked by intermolecular contacts, $\mathrm{O} 2-\mathrm{H} 2 \cdots \mathrm{~S} 1^{\mathrm{i}}$ and $\mathrm{C} 5-\mathrm{H} 5 \cdots \mathrm{O} 3^{\mathrm{ii}}$ (see Table 2 for symmetry codes), forming a one-dimensional chain parallel to the $b$ axis (Fig. 2).

## Experimental

A solution of 2-aminopropionic acid ( $4.45 \mathrm{~g}, 0.05 \mathrm{~mol}$ ) in acetone ( 20 ml ) was added dropwise to an acetone solution ( 20 ml ) containing an equimolar amount of benzoyl isothiocyanate in a two-necked round-bottomed flask. The solution was refluxed for about 5 h and then filtered into a beaker containing ice. The yellow precipitate was washed with cold acetone-distilled water, before being dried and kept in a desiccator (yield 10.72 g , $85 \%$, m.p. 430-431 K). Recrystallization from acetonitrile yielded single crystals suitable for X-ray analysis.

## Crystal data

$\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$
$M_{r}=252.29$
Triclinic, $P \overline{1}$
$a=7.3570(18) \AA$
$b=8.083(2) \AA$
$c=10.706(3) \AA$
$\alpha=87.395(4)^{\circ}$
$\beta=77.128(4)^{\circ}$
$\gamma=72.065(5)^{\circ} \AA^{\circ}$
$V=590.3(3) \AA^{3}$

## $Z=2$

$D_{x}=1.419 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 835 reflections
$\theta=1.9-25.0^{\circ}$
$\mu=0.27 \mathrm{~mm}^{-1}$
$T=298$ (2) K
Block, yellow
$0.49 \times 0.33 \times 0.31 \mathrm{~mm}$

## Data collection

Bruker SMART APEX CCD areadetector diffractometer
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Bruker, 2000)
$T_{\text {min }}=0.878, T_{\text {max }}=0.920$
5591 measured reflections

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0479 P)^{2}\right. \\
& \quad+0.2021 P] \\
& \quad \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.21 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.037$
$w R\left(F^{2}\right)=0.104$
$S=1.11$
2076 reflections
156 parameters
H -atom parameters constrained
1.385 (2) 1.385 (2)
1.315 (2)
1.454 (2)

| $\mathrm{C} 8-\mathrm{N} 1-\mathrm{C} 7$ | $128.30(16)$ | $\mathrm{N} 2-\mathrm{C} 8-\mathrm{S} 1$ | $123.69(14)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 8-\mathrm{N} 2-\mathrm{C} 9$ | $122.46(15)$ | $\mathrm{N} 1-\mathrm{C} 8-\mathrm{S} 1$ | $118.56(14)$ |
| $\mathrm{N} 2-\mathrm{C} 8-\mathrm{N} 1$ | $117.74(16)$ |  |  |

Table 2
Hydrogen-bond geometry ( $\AA{ }^{\circ},^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| C9-H9 . .S1 | 0.98 | 2.67 | 3.033 (2) | 102 |
| $\mathrm{N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 1$ | 0.86 | 2.01 | 2.669 (2) | 133 |
| $\mathrm{N} 2-\mathrm{H} 2 A \cdots \mathrm{O}$ | 0.86 | 2.39 | 2.698 (2) | 102 |
| $\mathrm{O} 2-\mathrm{H} 2 \cdots \mathrm{~S} 1^{\text {i }}$ | 0.82 | 2.32 | 3.1190 (16) | 166 |
| $\mathrm{C} 5-\mathrm{H} 5 \cdots \mathrm{O}^{\text {ii }}$ | 0.93 | 2.30 | 3.180 (3) | 158 |

Symmetry codes: (i) $x, y+1, z$; (ii) $x, y-1, z$.

H atoms were positioned geometrically $[\mathrm{O}-\mathrm{H}=0.82, \mathrm{~N}-\mathrm{H}=0.86$ and $\mathrm{C}-\mathrm{H}=0.93$ or $0.96 \AA$ (methyl)] and constrained to ride on their parent atoms, with $U_{\text {iso }}(\mathrm{H})=1.2$ (1.5 methyl and hydroxyl) $U_{\mathrm{eq}}(\mathrm{C} / \mathrm{N} / \mathrm{O})$.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON ( Spek, 2003).

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