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Key indicators

Single-crystal X-ray study T = 298 KMean σ (C–C) = 0.003 Å R factor = 0.037 wR factor = 0.104 Data-to-parameter ratio = 13.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(3-Benzoylthioureido)propionic acid

The molecular structure of the title compound, $C_{11}H_{12}N_2O_3S$, adopts a *cis–trans* configuration with respect to the positions of the propionic acid and benzoyl groups relative to the S atom across the thiourea C–N bonds. In the crystal structure, molecules are linked by O–H···S and C–H···O interactions, forming a one-dimensional chain parallel to the *b* axis.

Comment

The title compound, (I), is isomorphous with 3-(3-benzoylthioureido)propionic acid (Yusof & Yamin, 2003); they differ in that the 3-benzoylthioureido group is attached to the β alanine group in the latter compound. The molecular structure adopts a *cis-trans* configuration with respect to the positions of the propionic acid and benzoyl groups relative to the S atom across the C8-N2 and C8-N1 bonds, respectively.



The bond lengths and angles (Table 1) are in normal ranges (Allen *et al.*,1987) and comparable with those in the β -alanine derivative. The central carbonylthiourea group (S1/C8/N1/N2/C7), and the phenyl (C1–C6) and ethanoic acid fragments are planar, with a maximum deviation of 0.037 (2) Å for atom C7. The central carbonylthiourea group makes dihedral angles



Figure 1

The molecular structure of the title compound, (I), with displacement ellipsoids drawn at the 50% probability level. Dashed lines indicate hydrogen bonds.

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Figure 2

Packing diagram of compound (I), viewed down the *c* axis. The dashed lines denote $O-H \cdots S$ and $C-H \cdots O$ hydrogen bonds.

with the phenyl and ethanoic acid fragments of 27.52 (9) and 24.20 $(12)^{\circ}$, respectively. The phenyl ring is inclined to the ethanoic acid fragment by 4.19 (4)°.

There are three intramolecular hydrogen bonds C9– H9...S1, N2–H2A...O1 and N2–H2A...O3 (Table 2) and, as a result, two pseudo-five-membered rings (C9–H9–S1– C8–N2), (N2–H2A–O3–C11–C9) and a pseudo-sixmembered ring (N2–H2A–O1–C7–N1–C8) are formed (Fig. 1). In the crystal structure, the molecules are linked by intermolecular contacts, O2–H2...S1ⁱ and C5–H5...O3ⁱⁱ (see Table 2 for symmetry codes), forming a one-dimensional chain parallel to the *b* axis (Fig. 2).

Experimental

A solution of 2-aminopropionic acid (4.45 g, 0.05 mol) in acetone (20 ml) was added dropwise to an acetone solution (20 ml) containing an equimolar amount of benzoyl isothiocyanate in a two-necked round-bottomed flask. The solution was refluxed for about 5 h and then filtered into a beaker containing ice. The yellow precipitate was washed with cold acetone–distilled water, before being dried and kept in a desiccator (yield 10.72 g, 85%, m.p. 430–431 K). Recrystallization from acetonitrile yielded single crystals suitable for X-ray analysis.

Crystal data

$C_{11}H_{12}N_2O_3S$	Z = 2
$M_r = 252.29$	$D_x = 1.419 \text{ Mg m}^{-3}$
Triclinic, P1	Mo $K\alpha$ radiation
a = 7.3570 (18) Å	Cell parameters from 835
b = 8.083 (2) Å	reflections
c = 10.706 (3) Å	$\theta = 1.9-25.0^{\circ}$
$\alpha = 87.395 \ (4)^{\circ}$	$\mu = 0.27 \text{ mm}^{-1}$
$\beta = 77.128 \ (4)^{\circ}$	T = 298 (2) K
$\gamma = 72.065 \ (5)^{\circ}$	Block, yellow
V = 590.3 (3) Å ³	$0.49 \times 0.33 \times 0.31 \text{ mm}$
Data collection	
Bruker SMART APEX CCD area-	2076 independent reflections
detector diffractometer	1858 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.020$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.0^{\circ}$
(SADABS; Bruker, 2000)	$h = -8 \rightarrow 8$
$T_{\min} = 0.878, T_{\max} = 0.920$	$k = -9 \rightarrow 9$
5591 measured reflections	$l = -12 \rightarrow 12$

Refinement

2 1 F

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0479P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	+ 0.2021P]
$\nu R(F^2) = 0.104$	where $P = (F_0^2 + 2F_c^2)/3$
= 1.11	$(\Delta/\sigma)_{\rm max} < 0.001$
076 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$
56 parameters	$\Delta \rho_{\rm min} = -0.18 \text{ e } \text{\AA}^{-3}$
I-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

S1-C8	1.6778 (19)	N1-C8	1.385 (2)
O1-C7	1.212 (2)	N1-C7	1.385 (2)
O2-C11	1.321 (2)	N2-C8	1.315 (2)
O3-C11	1.195 (2)	N2-C9	1.454 (2)
C8-N1-C7	128.30 (16)	N2-C8-S1	123.69 (14)
C8-N2-C9	122.46 (15)	N1-C8-S1	118.56 (14)
N2-C8-N1	117.74 (16)		

Table 2		
Hydrogen-bond geometry ((Å,	°).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C9-H9\cdots S1 N2-H2A\cdots O1 N2-H2A\cdots O3 O2-H2\cdots S1^{i} C5-H5\cdots O3^{ii}$	0.98	2.67	3.033 (2)	102
	0.86	2.01	2.669 (2)	133
	0.86	2.39	2.698 (2)	102
	0.82	2.32	3.1190 (16)	166
	0.93	2.30	3.180 (3)	158

Symmetry codes: (i) x, y + 1, z; (ii) x, y - 1, z.

H atoms were positioned geometrically [O-H = 0.82, N-H = 0.86and C-H = 0.93 or 0.96 Å (methyl)] and constrained to ride on their parent atoms, with $U_{iso}(H) = 1.2$ (1.5 methyl and hydroxyl) $U_{eq}(C/N/O)$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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